

Supporting Information

Preparation of Graphene

To a dispersion of GO (100 mg) in water (50 mL), citric acid and poly(ethylene glycol) (PEG) mixed in a 1:3 molar ratio (0.25 g citric acid, 1.56 g PEG) in 20 mL of water were added. The solution was placed in a 100 mL Teflon lined autoclave and heated in a vacuum oven at 180 °C for 26 h. The resulting Graphene was filtered and washed with water, dried and stored in an inert atmosphere. The weight of the resulting product (Neat Graphene) is 33.2 % of the total weight of the MoO₂/Graphene composite and therefore in the composite, the MoO₂:Graphene proportion is 2:1. Similarly, two more composites with different MoO₂:Graphene proportions were also prepared for optimizing the MoO₂:Graphene proportion. When 70 mg and 130 mg of GO were used (all other contributions remaining the same), neat Graphene was found to be 21% and 40.5% of the total weight of the MoO₂/Graphene composite. So three MoO₂/Graphene composites with MoO₂:Graphene in 2:1, 3.76:1 and 1.45:1 ratios were prepared.

Preparation of MoO₂

To a dispersion 0.558 mM ammonium heptamolybdate ((NH₄)₆Mo₇O₂₄ .4H₂O) in water (50 mL), citric acid and poly(ethylene glycol) (PEG) mixed in a 1:3 molar ratio (0.25 g citric acid, 1.56 g PEG) in 20 mL were added. The solution was placed in a 100 mL Teflon lined autoclave and heated in a vacuum oven at 180 °C for 26 h. The resulting product designated as neat MoO₂ was filtered and washed with water, dried and stored in an inert atmosphere.

Preparation of MoO₂/Graphene oxide

To a dispersion of GO (100 mg) in water (50 mL), 0.558 mM ammonium heptamolybdate ((NH₄)₆Mo₇O₂₄ .4H₂O) in water (20 mL) and 1.56 g of poly(ethylene glycol) (PEG) were added.

The solution was placed in a 100 mL Teflon lined autoclave and heated in a vacuum oven at 180 °C for 26 h. The resulting product labeled as MoO₂/Graphene oxide composite was filtered and washed with water, dried and stored in an inert atmosphere.

Calculation of theoretical specific capacity of MoO₂/Graphene composite:

- Theoretical specific capacity of MoO₂ = 838 mAh g⁻¹ (for 4 Li⁺ insertion/deinsertion)
- Theoretical specific capacity of Graphene = 623 mAh g⁻¹

$$\begin{aligned} \text{Theoretical specific capacity of MoO}_2/\text{Graphene} &= 66.8 \% \text{ MoO}_2 + 33.2 \% \text{ Graphene} \\ &= 66.8 \% * 838 + 33.2 \% * 623 \\ &= 766.8 \text{ mAh g}^{-1} \end{aligned}$$

The galvanostatic charge-discharge characteristics of the MoO₂/Graphene composites with MoO₂:Graphene weight ratios of 2:1, 3.76:1 and 1.45:1 are shown below in the Fig. S1. The charge-discharge curves recorded at a current density of 540 mA g⁻¹ (Fig. S1a) for the first cycle of MoO₂/Graphene in 2:1, 3.76:1 and 1.45:1 proportions show charge/discharge capacities, respectively, of 703/1455, 535/988 and 552/1094 mAh g⁻¹. As can be observed from Fig. S1a, the MoO₂/Graphene composite (2:1) shows a high initial reversible capacity of 700 mAh g⁻¹ and the MoO₂/Graphene composite (3.76:1) shows an initial capacity of 535 mAh g⁻¹. The low graphene content in the MoO₂/Graphene composite (3.76:1), renders the material less conducting which is responsible for the observed poor capacity. On the other hand, in the MoO₂/Graphene composite (1.45:1), the active material (MoO₂) concentration is low, which can cause the re-stacking of graphene layers, and therefore a low capacity (552 mAh g⁻¹) is observed. It is obvious that the MoO₂/Graphene composite (2:1), has an optimal balance of the two components, which

results in good electrochemical performance. After 50 cycles the MoO₂/Graphene composites in 2:1, 3.76:1 and 1.45:1 shows reversible capacities of 760, 575 and 320 mAh g⁻¹ (Fig. S1b).

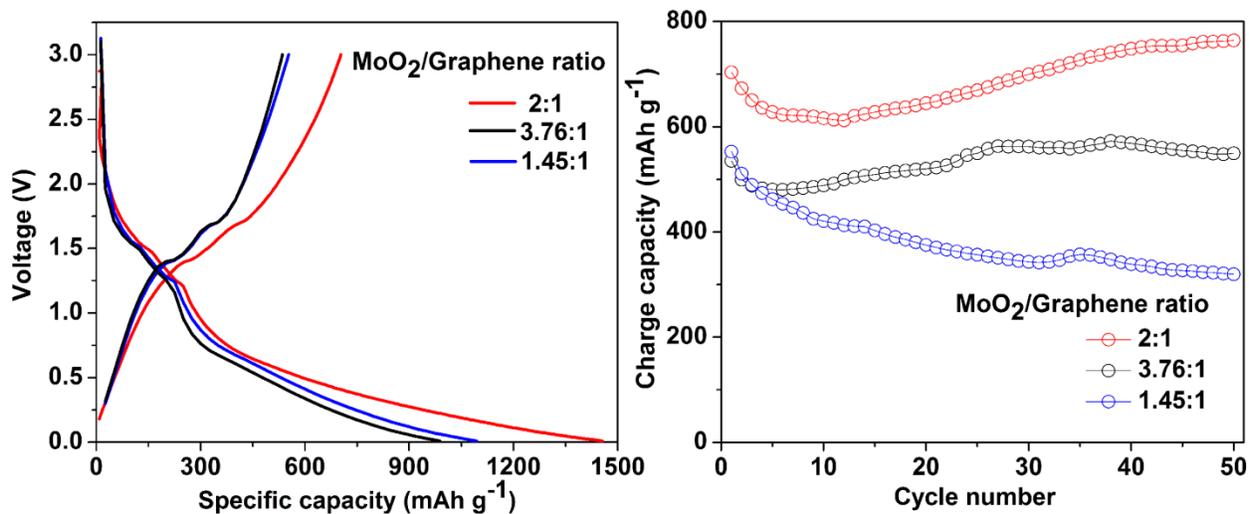


Fig. S1. (a) Galvanostatic charge-discharge curves of MoO₂/Graphene composite with different proportions. (b) cyclability of the composites; all curves recorded at a current density of 540 mA g⁻¹.

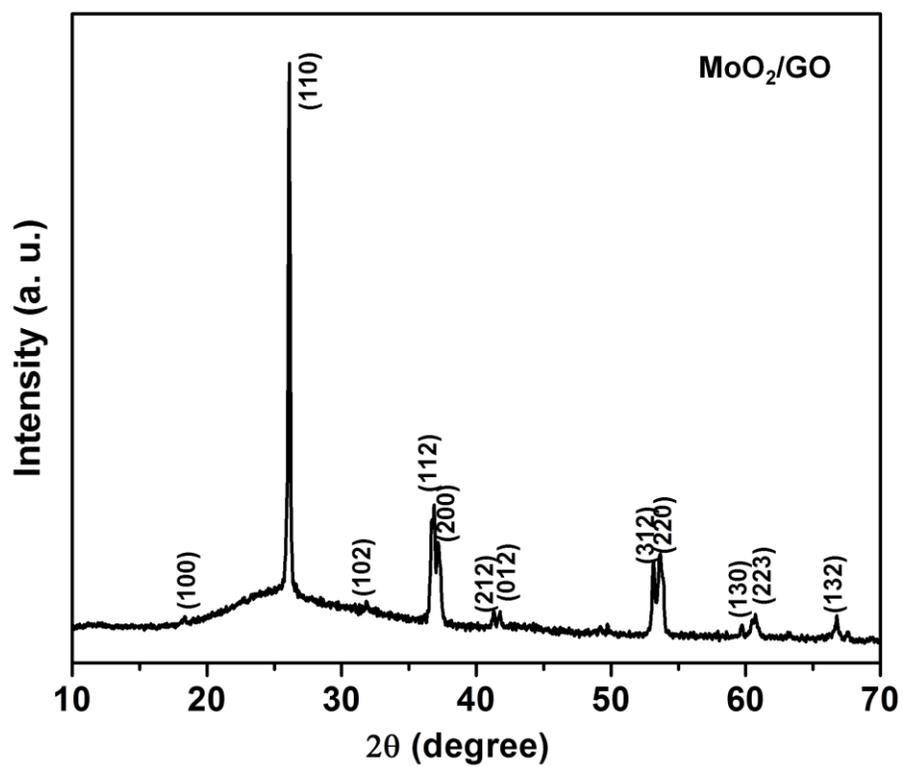


Fig. S2. XRD pattern of MoO₂/Graphene oxide composite

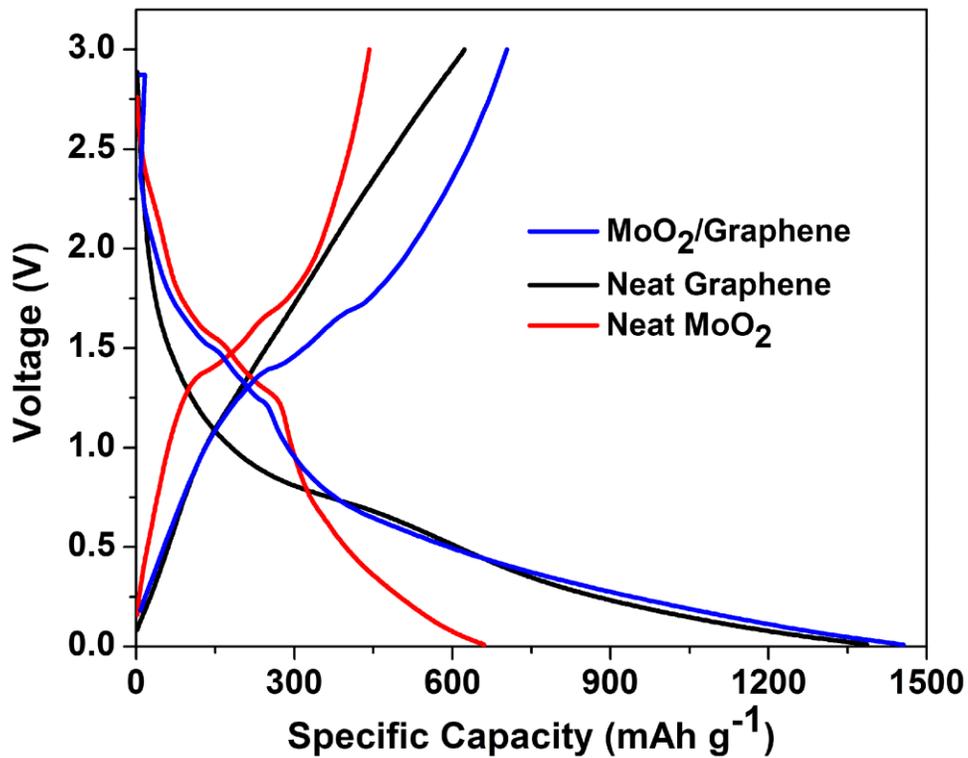


Fig. S3. Galvanostatic charge-discharge curves for the first cycle of Neat MoO₂, Neat Graphene and MoO₂/Graphene composite at current densities of 200, 200 and 540 mA g⁻¹, respectively.

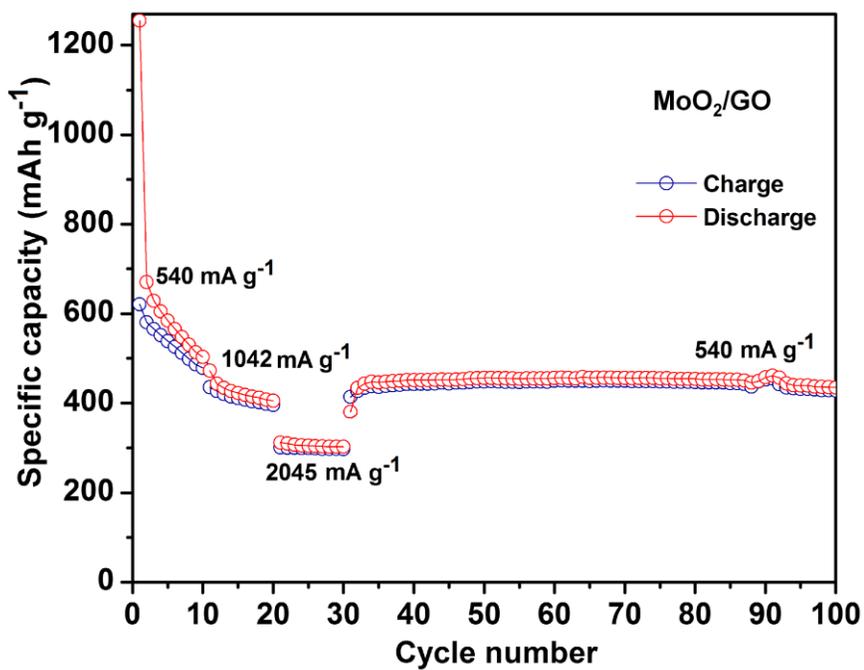


Fig. S4. Rate capability of MoO₂/GO composite.

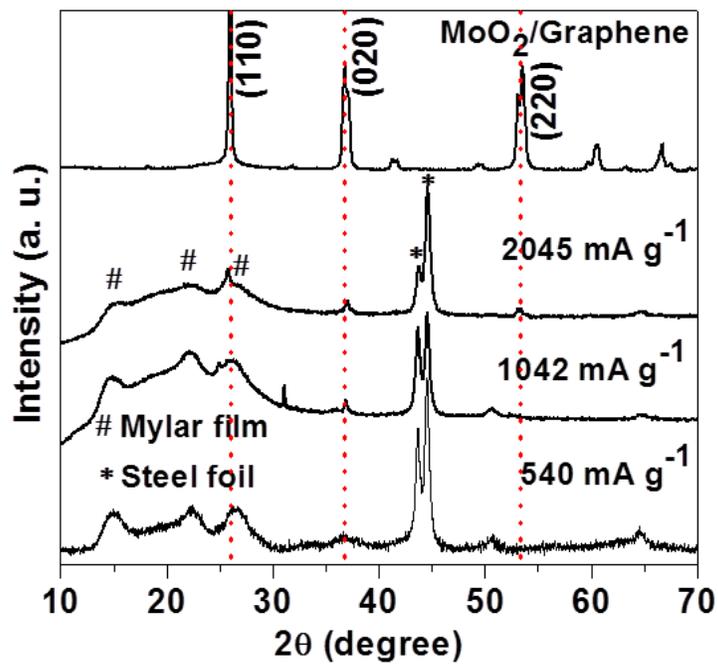


Fig. S5. *Ex-situ* XRD patterns of MoO₂/Graphene electrode after subjecting it to 50 charge–discharge cycles, at different C-rates of 540, 1042 and 2045 mA g⁻¹.